**CHROMBIO. 2 27** 

# **ANALYSIS OF FATTY ACIDS FROM HUMAN LIPIDS BY GAS CHROMATOGRAPHY**

# **G-J. HAAN', S. VAN DER HEIDE and B.G. WOLTHERS\*\***

*University Hospifal Groningen, Central Laboratory for Clinical Chemistry, dostersingel 59, Groningen (The Netherlands)* 

**(First received April 5th, 1978; revised manuscript received June 27th, 1978)** 

#### **SUMMARY**

**A rapid, quantitative method is described for the analysis of fatty acids from human lipids, namely serum lipids and lipids from adipose tissue biopsies. The method includes extraction of serum lipids with chloroform-methanol, hydrolysis with tetramethylammonium hydroxide, methylation with methyl iodide and N,N-dimethylformamide and gas chromatographic analysis on a Supelcoport SP-2330 column. Fat biopsies are analysed without estraction Optimal hydrolysis conditions have been investigated.** 

#### **INTRODUCTION**

**Many procedures have been described for the quantitative analysis of fatty acids [l]** \_ **All these procedures are trmeconsuming with regard to extraction, hydrolysis and esterification for gas chromatography (GC). Because of the interest at our hospital in fatty acids of adipose tissue from diabetic children and of serum lipids from children with cystic fibrosis and also from adults, as a part of a population screening programme, we searched for more rapid procedures. Greeley 123 described a new method for alkylation of acidic substances for GC and West 133 modified this method to prepare methyl esters from glycerides, alkyl paint resins and ester plasticizers. We tried this method on our samples and modified it to obtain rapid hydrolysis and methylation of the lipids under mild conditions. Combined with a simple extraction procedure and an appropriate GC procedure, an efficient analysis of fatty acids was achieved for clinical chemical purposes.** 

**<sup>\*</sup>Present address: Independent Regional Laboratories for Clinical Analysis, Nieuwe Gracht 39, Haarlem and Watercirkel302, Amstelveen, The Netherlands.** 

**<sup>\*\*</sup>To whom correspondence should be addressed.** 

#### **MATERIALS AND METHODS**

# *Chemicals*

All chemicals were of analytical grade and were purchased from Merck (Darmstadt, G.F.R.) except the following chemicals: fatty acids and methyl esters of the fatty acids were from Applied Science Labs., (State College, Pa., U.S.A.), tetramethylammonium hydroxide, 24% in methanol, was from Fluka (Suchs, Switzerland), trilinolein was from ICN Pharmaceuticals (Plainview, N.Y., U.S.A.), and  $10\%$  SP-2330 on Chromosorb W AW  $(100-120 \text{ mesh})$  was from Supelco (Bellefonte, Pa., U.S.A.). High-performance thin-layer chromatography plates, silica gel 60,  $10 \times 20$  cm, were from Merck.

### *Procedures*

Standard concentrations of fatty acids and their methyl esters were made by dissolving 15-20 mg of each lipid *in* 100 ml of chloroform. The following fatty acids and their methyl esters were used: myristic acid  $(C_{14})$ , pentadecanoic acid (C<sub>15</sub>), palmitic acid (C<sub>16:0</sub>), palmitoleic acid (C<sub>16:1</sub>), stearic acid  $(C_{18:0})$ , oleic acid  $(C_{18:1})$ , linoleic acid  $(C_{18:2})$  and arachidonic acid  $(C_{20:4})$ . Pentadecanoic acid, 130 mg dissolved in 100 ml of chloroform, was used as an internal standard\_

To 0.5 ml serum, 0.2 ml of internal standard solution was added in a reagent tube. For the extraction of serum Iipids and the free fatty acids, 1.5 ml of water, acidified with concentrated HCl to pH 2, and 5 ml of chloroform-methanol  $(7:3, v/v)$  were added to the mixture of serum and internal standard. The tubes were shaken twice for about 30 sec on a vortex mixer. After centrifugation for 5 min at 700 g, the chloroform layer was removed and the extraction was repeated with 3 ml of chloroform-methanol\_ Both the chloroform layers were combined. The chloroform extract was evaporated under a stream of nitrogen at 40" to complete dryness.

For the saponification of the Iipids 0.1 ml of tetramethylammonium hydroxide solution and 0.1 ml of methanol were added. The tubes were tightly capped and were allowed to stand for 10 min at 70" in a heating block. After cooling to room temperature the methyl esters were prepared by adding 1 ml of N,N-dimethylformamide and 0.1 ml of methyl iodide. The tubes were shaken and the precipitate was removed by centrifugation. The supematant was ready for injection into the gas chromatograph. Fat biopsies were weighed and then saponified and methylated as described above.

The gas chromatograph used was a Packard-Becker Model 428 equipped with flame ionization detectors. A 5- $\mu$ l aliquot of the sample was injected on a 2-m glass column. (2 mm I.D.) packed with 10% SP-2330 on Chromosorb W AW, 100-120 mesh. The carrier gas was nitrogen, flow-rate 20 ml/min. The injection port temperature was 220° and the detector temperature 250°. The oven temperature was programmed to run for 5 min isothermally at  $130^{\circ}$ , followed by a temperature rise of  $5^{\circ}/$ min to  $220^{\circ}$ .

Methyl ester standards in chloroform of all the *acids* were injected before and after each series of sample analyses. When differences of less than 5% in the peak heights of all the fatty acids, relative to the internal standard (IS.), were observed between the first and the last standard run, the first standard run was

**used for calculating the amounts of the individual fatty acids (FA) in terms of unmethylated acids. Calculations were made according to the equation** 

 $\mu$ g FA per sample = peak height  $FA \times \mu g$  I.S. **X factor peak height IS.** 

**where factor (standard) is** 

 $\mu$ g FA $\times$  peak height I.S. peak height  $FA \times \mu g$  I.S.

**Thin-layer chromatography was performed according to the method of Van Gent [ 41 using high-performance silica-gel plates with consecutive development in chloroform and hesane-chloroform (3:l). The slides were sprayed with the charring agent, 10% sulphuric acid in acetone-water (1:l).** 

**NMR spectra were made on a Bucker spectrometer Model HX-360 (360 MHz). Further details are given in the figures. Mass spectra were made on a**  Varian gas chromatograph-mass spectrometer combination Model MAT 112, **on a 2-m column (1.2 mm I.D.) packed with 10% SP-2330 on Chromosorb W AW, 100-120 mesh. The carrier gas was helium, flow-rate 9 ml/min. The oven temperature was 190", separator temperature 250", injection port temperature 250" and the source temperature 200". Tine ionization energy was 70 eV and the scan speed 200 a.m.u./sec.** 

**Methylation with diazomethane as a reagent was performed according to the procedure described by Schenk and Gellerman [ 5]\_** 

#### **RESULTS**

**The methy!ation procedure was checked by methylating a fatty-acid mixture of known composition in chloroform (see Table I). Of this mixture 1 ml was evaporated under a stream of nitrogen at 40". To the dry fatty acids, all the saponification-methylation reagents were added as described above, omitting**  heating for saponification. The  $5 \mu l$  of the sample were injected, followed by  $5 \mu l$  of the standard mixture of methyl esters of the fatty acids. To see wheth**er the methylation was complete the amount of each acid, including pentidecanoic acid, was calculated, considering pentadecanoic acid not as an internal standard\_ The methylation was also compared to the procedure using diazomethane as reagent. As can be seen in Table I, the methylation is almost complete for all fatty acids, and gave results comparable with the diazomethane** \_ **procedure\_** 

**The original method of West [S] described the saponification at 100". Because loss of the polyunsaturated fatty acids is possible at this temperature, we performed the same procedure with the fatty acid mixture at different saponification temperatures to see whether temperature effects were present. The re, sults are given in Table II. Recoveries of both polyunsaturated acids C18:2 and**   $C_{20:4}$  are temperature dependent. Performing the procedure with linoleic acid **at high saponification temperature resulted in the appearance of two new peaks in the gas chromatograms immediately after the partly disappeared linoleic acid peak. Mass spectrometric analysis of these peaks revealed that they were due to compounds with the structure and molecular weight of methyl lino-** 

#### **TABLE I**

## **METHYLATION OF A STANDARD MIXTURE OF FATTY ACIDS WITH THE REPORT-ED PROCEDURE AND WITH DIAZOMETHANE AS METHYLATING REAGENT**

**Figures are the mean recovery values for** 24 **estimates (duplicate injections of three samples on four different days), expressed in pg per sample and as percentages of the original standard values\_ Recoveries were calculated by dividing average peak heights of the methylated free fatty-acid standard by those of the average peak heights of methyl ester standard\_** 



# **TABLE II**

 $\mathcal{F}_{\mathcal{F}_{\mathbf{a}}}$ 

#### **EFFECT OF DIFFERENT SAPONIFICATION TEMPERATURES ON THE FATTY-ACID STANDARD MIXTURE**

Figures are the mean values for 6 estimates, expressed in  $\mu$ g per sample and as percentages **(in parentheses) of the original standard values.** 

**.\_\_\_.\_\_ \_\_\_** 



**264** 

**leate (Fig. 1). NMR spectral analysis showed that these compounds were the result of migration of the double bonds to conjugated positions during the saponification step (Fig. 2). The double bonds are probably located in the 9-10-cis, 11-12-trans and 10-11-trans, 12-13-cis positions instead of the 9-10-k, 12-13ck positions, as in linoleic acid. We expect that a similar process occurs during saponification of arachidonic acid, at an even faster rate (Table II). Losses by oxidative processes during the analytical steps were not observed because the sum of the amounts of the**  $C_{18:2}$  **acids was the same as the original amount of linoleic acid. To prevent the migration of double bonds, saponification at lower temperatures, where losses of linoleic acid and arachidonic acid are minimal, is necessary\_ To test the saponification at lower temperatures, trilinolein was used. Trilinolein was dissolved in chloroform (19.9 mg/lOO ml) with pentadecanoic acid as an internal standard (15.3 mg/lOO ml). Of this solution, 0.5, 1, 2 and 3 ml were evaporated under a stream of nitrogen at 40". Saponification was carried out as described above at 70" for 10 min. Under these conditions the saponification is almost complete with no losses of linoleic acid (Table III) \_** 

**To determine whether hydrolysis of serum lipids is complete under these circumstances, aliquots of pooled serum were analyzed, using different tempera-**





**Fig. 1. Parts of the mass spectra of: (A) methyl linoleate and (B) the formed compounds.** 



- .

电重用光线 医内皮囊 医动物

 $\overline{1}$ 

**tures and- incubation times during the saponification step. The results** are shown in Table IV. Silica-gel thin-layer chromatography [4] revealed that saponifica**tion of the sample at 100" was complete after 30 min because** no fatty-acid **esters were visible in the chromatogram (Fig. 3).** 

**Moreover, the presence of cholesterol in** the serum samples is cIearly shown, as **should be expected. However, its presence does not interfere in the GC analysis. This was proved by injecting a solution of free cholesterol. After 1 h,** 

**TABLE III** 

### **SAPONIFICATION OF TRILINOLEIN FOR 10 MIN AT 70"**

Figures are the mean recoveries of duplicate injections of duplicate samples expressed in  $\mu$ g **linoleic acid per sample and in percentages of the amount present.** 

**\_\_\_\_-** 



## **TABLE IV**

**EFFECT OF TEMPERATURE AND TIME ON THE SAPONIFICATION OF SERUM LIPIDS** 

**Pooled serum (0.5 ml) was used according to the procedure described in the text, including internal standard. Figures are the mean values of duplicate injections of duplicate samples expressed as percentages of the highest yield found.** 





Fig. 3. Thin-layer chromatography on silica gel of a serum lipid extract in chloroform (1), **and of serum extracts saponified for 10 min at 70" (2), for 90 min at 60" (3) and for 30 min at 100" (4). CHE = cholesterol esters; MFA = methyl fatty acids; TG = triglycerides: CHOL = cholesterol; FFA = free fatty acids; PPL = phospholipids.** 

# **TABLE V**

## **EXTRACTION RECOVERY OF FATTY ACIDS AND TRINOLEIN ADDED TO SERUM**





**keeping the oven at 220", no peak was observed\_ Most likely cholesterol has a very large retention time at this temperature or is completely retained on the column\_ The samples saponified at 70" and 60" still contained traces of cholesterol esters and phospholipids. Triglycerides were not seen. As can be seen in Table IV, no ideal conditions, in which complete hydrolysis and no loss of polyunsaturated acids occur together, can be found. However, results for 10 min at 70" and 90 min at 60" are quite satisfactory.** 

**The extraction procedure was checked by adding fatty acids and trilinolein to serum and determining the recovery\_ Dried aliquots (1 ml) of both a fattyacid standard mixture and trilinolein were determined without the extraction step and also after dissolving them in 0.5 ml serum, followed by extraction\_ Recoveries are shown in Table V.** 

**The reproducibility of the method was tested by analysing pooled serum ten times according to the above procedure (Table VI).** 

**The fatty-acid composition of the pooled sera of 40 healthy laboratory technicians (aged 18-55 years) together with some values from the literature are listed in Table VII.** 

**Some examples of the usefulness of the method for clinical purposes are given in Tables VIII and IX. Table VIII shows the composition of the serum** 

# **TABLE VI**

**REPRODUCIBILITY OF TEN ANALYSES OF THE SAME SERUM** 



## **TABLE VII**

**FATTY-ACID COMPOSITION OF SERUM LIPIDS OF THE POOLED SERA OF 40 LABORATORY TECHNICIANS COMPARED WITH SOME VALUES FROM THE LITERATURE** 

**Figures represent percentages of the total amount of the fatty acids considered.** 



# **TABLE VIII**

# **FATTY-ACID COMPOSITION OF SERA OF HEALTHY CHILDREN COMPARED WITH CHILDREN SUFFERING FROM CYSTIC FIBROSIS**

**Figures are expressed as percentages of the total amount of the fatty acids considered.** 



# **TABLE IX**

.-

**FATTY-ACID COMPOSITION OF FAT BIOPSIES OF DIABETIC CHILDREN COM-PARED WITH VALUES FROM THE LITERATURE** 

 $C_{1.4}$  c<sub>16:0</sub> c<sub>16:1</sub> c<sub>18:0</sub> c<sub>18:1</sub> c<sub>18:2</sub> c<sub>20:4</sub> Others **Diabetic children, (n=23) % 2.8 20.9 6.7 4.0 46.7 15.8 2.3 -**  S.D. 0.77 2.09 1.40 1.24 2.58 3.31 0.82 **Rei 11 % 2.7 24.0 5.0 8.4 46.9 10.2 - 2.5** 

**Figures are expressed as percentages of the total amount of the fatty acids considered.** 

**fatty acids of children with cystic fibrosis compared with healthy children\_ Our results are in agreement with the results of others [9, 10],** *i.e.***, higher values for the monounsaturated and lower values for the polyunsaturated acids. The mean values of the fatty acid composition of fat biopsies compared with literature values are given in Table IX.** 

## **DISCUSSION**

**The method presented in this paper has many advantages in comparison with other procedures. The extraction is simple because no further extractions to purify the samples from non-lipid material are necessary\_ Saponification for 10 min & 70" and the subsequent methylation are very simple and rapid and**  **occur under mild circumstances without oxidation or double-bond migration in the polyunsaturated acids. However, the saponification can not be totally complete without loss of polyunsaturated acids, but the same probably holds true for other saponification procedures. For practical reasons we have taken a saponification procedure of 10 min at 70", which will give about 85% saponification, although about 92% saponification can be achieved with the conditions 90 min at 60".** 

**Reproducibility is good and the results of determinations of the fatty-acid composition of 40 laboratory technicians agree fairly well with values found in the literature\_ Any differences could be due to dietary circumstances. Furthermore, the method is very well suited to the determination of the fatty-acid composition of fat biopsies\_** 

**In a laboratory with good automated equipment for GC, about. 50 samples can be analysed daily by one technician. This method should also be suitable for the assay of other fatty acids and for the analysis of other types of sample.** 

# **ACKNOWLEDGEMENTS**

**The authors wish to express their gratitude to Dr. N.M. Drayer, Drs. A. Temmerman and Dr. K. Knol of the Department of Pediatrics, and to Drs. E.W. Kwarts and Dr. A. Groen of the Central Laboratory for their cooperation throughout the investigations; to Dr. R. Kaptein and Mr. K. Dijkstra of the PhysicaI Chemistry Laboratory, University of Groningen, for making the NMR spectra, to Dr. J. Bus of the Unilever Research Laboratories, Vlaardingen, for his help in the interpretation of the NMR spectra, to Mr. J.J. Hoks for drawing the figures and to Drs. J.J. Pratt for reading the manuscript.** 

#### **REFERENCES**

- **1 A. Kuksis, J. Chromatogr., 143 (1977) 3.**
- **2 R.H. Greeley, J. Chromatogr., 88 (1974) 229.**
- **3 J.C. West, Anal. Chem., 47 (1975) 1708.**
- **4 C.M. van Gent, 2. Anal. Chem., 236 (1968) 344.**
- **5 H. Schenk and J.L. Gellerman, Anal. Chem., 32 (1960) 1412.**
- **6 Wissenschaftliche Tabellen, Geigy S.A., Basle, Switzerland, 7th ed., 1968, p\_ 600.**
- **7 W.C. Love, A. Cashell, M. Reynolds and N. Callaghan, Brit. Med. J., III (1974) 18.**
- **8 H. Wagener, in G. Schettler (Editor), Lipids and Lipidoses, Springer Verlag, Berlin, 1967, p\_ 197.**
- **9 M.L. Rosenlund, J.A. Selekman, H.K. Kim and D. Kritchevsky, Pediatrics, 59 (1977) 428.**
- **10 B.A. Underwood, CR. Denning and M. Navab, Ann. N.Y. Acad. Sci., 203 (1972) 237.**
- **11 R.C. Weast (Editor), Handbook of Chemistry and Physics, CRC Press, Ohio, 49th ed., 1968, p. D-190.**